

Anatomy of The Respiratory System & Partial Pressures

(Text Pg 120 – 124)

The respiratory system is composed of structures that allow the **passage of air** from outside the body to the lungs as well as structures that allow for **gas exchange** to occur.

3 Main Functions of the Respiratory System

1. Supply O₂
2. Remove CO₂
3. Regulate blood pH

The structures of the respiratory system are divided into two main zones:

A) Conductive Zone:

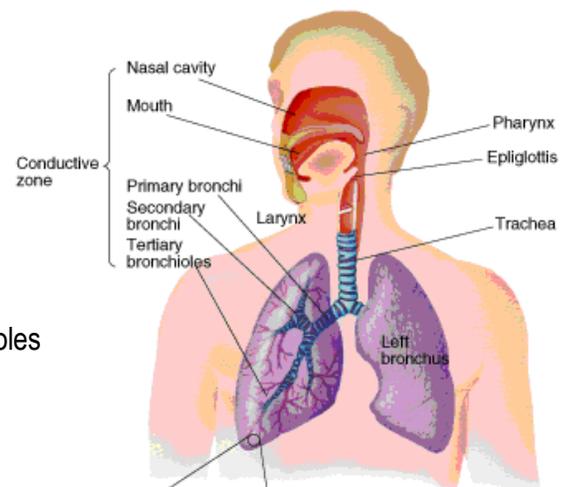
The combination of all the structures that convey air from the outside of the body to the lungs

Purpose:

- Transport air to the respiratory zone
 - **Filter** incoming air
 - **Humidify** incoming air
 - **Warm** the inspired air

Structures include:

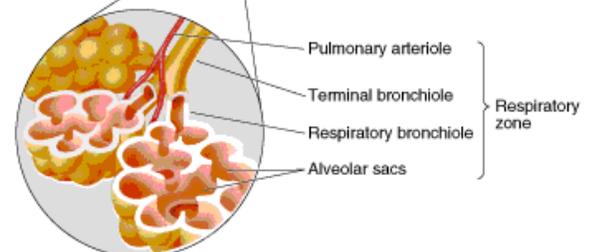
- Mouth & Nose
- Pharynx
- Larynx
- Trachea
- Bronchi
- Bronchioles



B) Respiratory Zone

The structures that are involved with the exchange of gases between inspired air and the blood

- Terminal Bronchiole
- Respiratory Bronchioles
- Alveolar ducts
- Alveoli



Purpose:

- Rapid Gas exchange (O₂ & CO₂) via diffusion
 - Alveoli = functional unit (1 cell thick membrane)
 - 300 million = ~ 2800 square feet of surface area
 - Alveoli are covered with capillaries (1 cell thick membrane)

DEMO: Holding Your Breath

- Lets Hold Our Breath for as long as we can!
 - How were you feeling when you had to stop? What makes us want to take our next breath?

Physiology of Respiration (Respiratory Control Centres)

- The main stimulus to breathe is the level of CO₂ in the blood. As CO₂ levels rise, the **Medulla Oblongata** (respiratory control center in the brain) triggers:
 - An increase in the breathing **Frequency (F)**
 - An increase in the **DEPTH** of each breath known as the **Tidal Volume (V_T)**
 - ✓ Both remain elevated until the level of CO₂ returns to acceptable levels.
- The **Pons** fine tunes our breathing rate (smoothes out the transition between inhalation and exhalation)

Atmospheric air is a mixture of three key gases

- Oxygen (O₂) = 20.93% pO₂ = 159.1 mmHg
- Carbon Dioxide (CO₂) = 0.03% pCO₂ = 0.23 mmHg
- Nitrogen (N₂) = 79.04% pN₂ = 600.7 mmHg

Three Rules That Determine Gas Movements For Breathing

1. Gasses move from high to low **pressure** (Law of partial pressures).
2. Gasses move from **high concentration** to **low concentration**.
3. Gasses move within the body by means of **diffusion**. (as the concentration gradient increases, greater rates of diffusion are observed)

Henry's Law

- The amount of gas that will dissolve/diffuse into a liquid is proportional to the partial pressure of the gas and the solubility of the gas.
- A gas will continue to dissolve into solution until equilibrium is reached.

The Partial Pressure of a Gas Determines Its Movement Between:

- The atmosphere and our lungs
- Our lungs and our blood
- Our blood and our cells
- As PO₂ ↓'s or PCO₂ ↑'s, it becomes harder to breath

What Happens When Altitude Changes?

- % of gas remains constant.
- The Partial pressures change.
 - o **Up** a mountain: Partial Pressure **drops** which limits O₂ delivery to the blood (e.g. PO₂ = 104 mmHg)
 - o **Below** sea level: Partial Pressure **rises** which increases O₂ delivery to the blood (e.g. PO₂ = 180 mmHg)

